Modifying WWTPs to Achieve Nitrogen Removal

April 26, 2017 Murfreesboro, TN

Wave Four Energy Management Workshop

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Organisms and Their Means of Respiration

- Aerobic use elemental oxygen
- ◆ Anoxic use nitrate (NO₃) or nitrite (NO₂)
- Anaerobic use other terminal electron acceptors (SO₄, CO₂) or none at all
- Facultative two or more means of respiration
- Fermentative no terminal electron acceptor

CBOD Removal

- Heterotrophic
- Aerobic, anoxic, anaerobic
- Floc formers

Nitrification

- Autotrophic
- Aerobic
- Not floc formers

Denitrification

- Heterotrophic
- Anoxic (facultative)

Denitrification Reactions

For one gram of NO₃-N that is denitrified:

2.47 g of methanol (~3.7 g of COD) are consumed

0.45 g of new cells are produced

3.57 g of alkalinity are formed

Denitrification: Biochemical Reactions

Sewage as carbon source:

$$C_{10}H_{19}O_3N + 10NO_3^- \rightarrow 5N_2 + 10CO_2 + 3H_2O + 10OH^- + NH_3$$

Factors Affecting Denitrification

- Substrate degradability
- pH
- Dissolved oxygen
- Temperature

Impact of DO on Denitrification Rates

| DO CONC, Mg/L |
|---------------|
|---------------|

Denitrification Rate

| 0.0 | 100% |
|-----|------|
| 0.1 | 40% |
| 0.2 | 20% |
| 0.3 | 10% |

Oxygen Savings with Denitrification

For every gram of NO₃-N that is reduced to nitrogen gas, 2.86 grams of oxygen are saved.

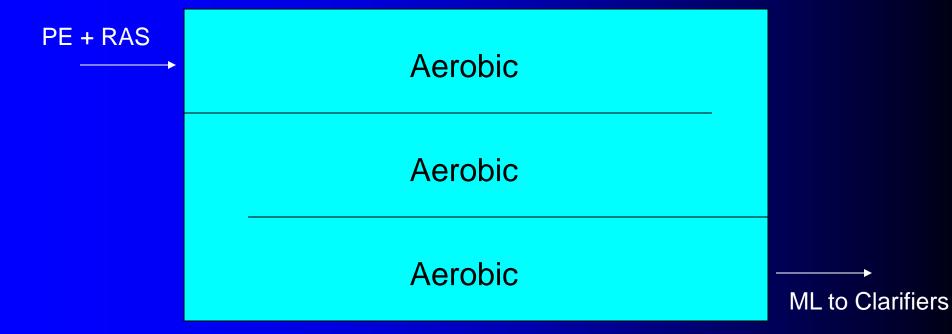
Performance of Single-Sludge Denitrification

- Can achieve high N removals (85% to 95%)
- Does not necessarily enhance sludge settleability in final clarifier
- Uses carbon source in influent
- Reduces the energy requirements for BOD removal from the wastewater (2.86 lb O₂ equivalent per lb of NO₃-N removed)
- About one-half of alkalinity required for nitrification is produced in anoxic zone

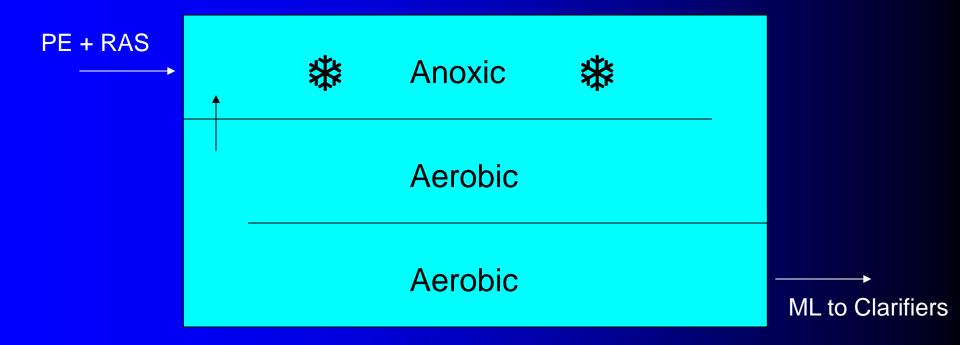
WWTP Changes to Achieve Nitrification-Denitrification

- Modify rectangular aeration basin with baffles to provide anoxic and aerobic zones
- Modify oxidation ditch to provide anoxic and aerobic zones
- Modify oxidation ditch operation with on/off aeration cycles to achieve denitrification
- Modify SBR system to include anoxic and aerobic cycles
- Modify step-feed system to include alternating anoxic and aerobic zones

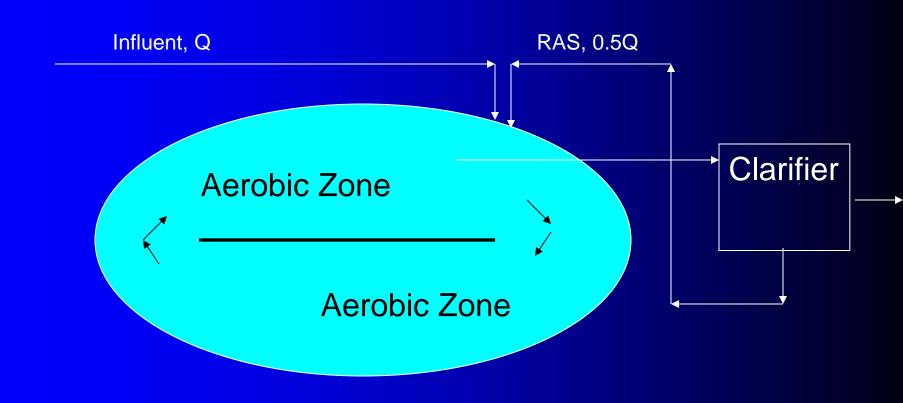
Rectangular Aeration Basin



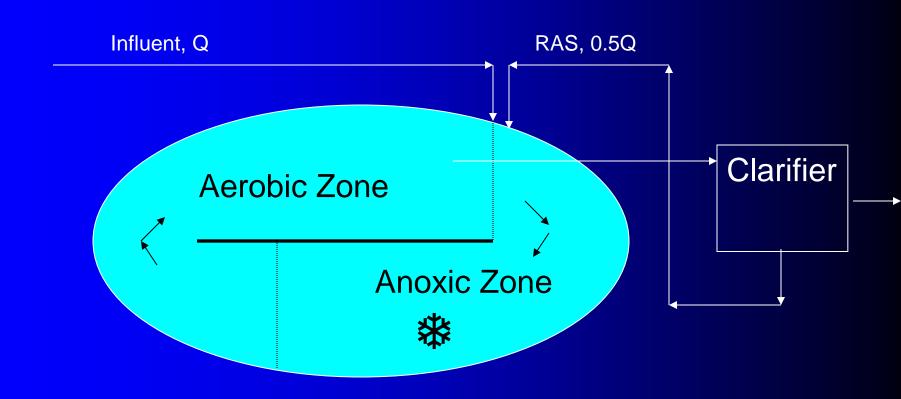
Modified Rectangular Aeration Basin



Oxidation Ditch Before Modification



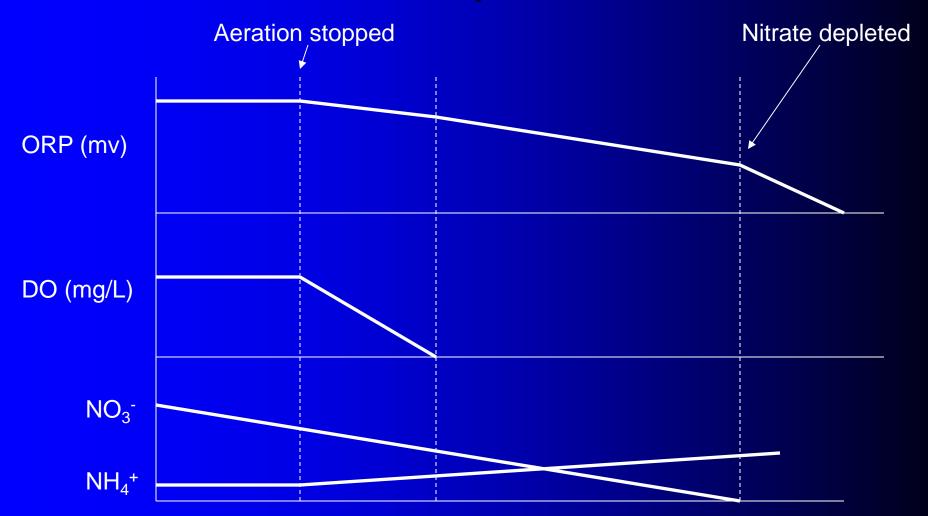
Oxidation Ditch After Modification



Intermittent Aeration for N Removal in Oxidation Ditch

- Cycle time for on/off operation of aerators may vary
- Process control with DO and ORP monitoring
- When aerator is off, must provide mixing
- During off period, oxidation ditch becomes anoxic reactor, and nitrate is consumed as bacteria degrade BOD
- ORP data are used to terminate off cycle and start aeration

Change in ORP and DO in On/Off Operation



Factors Affecting On/Off Operation

- Oxidation ditch HRT
- Influent flow rate
- TKN and BOD concentrations
- Number of on/off cycles per day
- Ditch MLSS concentration

Nitrogen Removal in Oxidation Ditch Using DO Control

Expected Effluent Quality:

BOD₅

TSS

Ammonia-N

NO_x-N

Total N

5 - 15 mg/L

10 - 20 mg/L

< 1 mg/L

5 - 10 mg/L

7 - 14 mg/L

Actual N Removal in Oxidation Ditches

| <u>WWTP</u> | <u>TKN</u> | <u>NH₃-N</u> | <u>NO</u> ₃ -N | N rem % |
|-------------|------------|-------------------------|---------------------------|---------|
| Α | 3.1 | 1.4 | 19.3 | 2 |
| В | 1.6 | 0.9 | 2.3 | 86 |
| С | 2.5 | 1.1 | 9.7 | 51 |
| D | 3.9 | 1.9 | 8.3 | 80 |
| E | 3.3 | 0.5 | 12.4 | 72 |

Nitrogen Removal in SBRs

- Use anoxic and aerobic cycles to effectively remove nitrogen
- Cycles are:
 - Fill (anoxic)
 - React (aerobic/anoxic)
 - Settle
 - Decant

Nitrogen Removal in SBRs

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BOD₅

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Ammonia-N

NO_x-N

Total N

5 - 15 mg/L

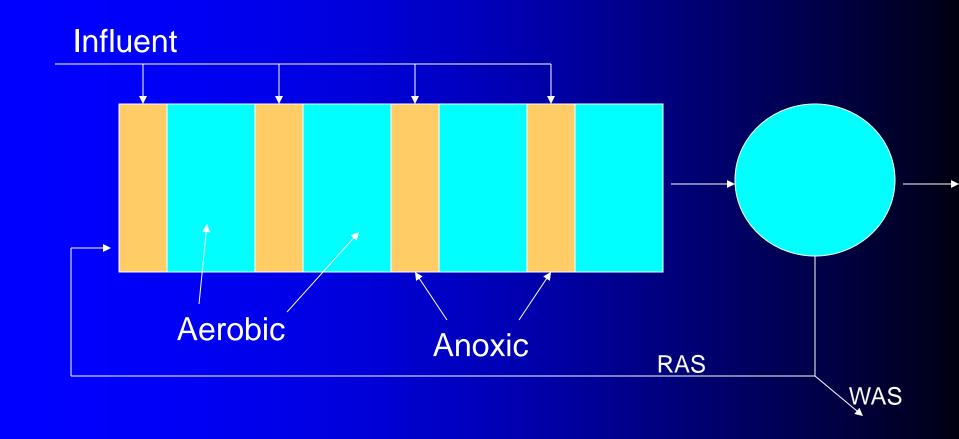
10 - 20 mg/L

< 1 mg/L

3 - 10 mg/L

5 - 12 mg/L

N Removal in Step-Feed Process



Conventional Activated Sludge



Add an Anoxic Zone using Baffle, Mixed Liquor Return, and Mixing



Before and After Effluent Quality

| | <u>Before</u> | <u>After</u> |
|--------------------|---------------|--------------|
| BOD ₅ | 5 - 25 mg/L | 5 - 15 mg/L |
| TSS | 10 - 25 mg/L | 10 - 20 mg/L |
| Ammonia-N | 1 - 5 mg/L | 1 - 2 mg/L |
| NO _x -N | 8 - 15 mg/L | 3 - 9 mg/L |
| Total N | 10 - 20 mg/L | 5 - 12 mg/L |
| *SVI | 125 - 225 | 50 – 125 |

^{*} impacts on mixed liquor at one facility

Principles in Implementing N Removal

- Consider a wide variety of alternatives
 - use on/off operation of aeration units
 - add anoxic zone and mixed liquor recirculat'n
 - optimize anoxic/aerobic periods
 - adjust SRT to enhance nitrogen removal
- Invest in plant-specific waste characterization data
- Engineer and operators should discuss changes
- Tailor to specific plant situation
- Balance risk and cost

Case Study: Pell City, AL WWTP

Single Oxidation Ditch Design flow rate (ave daily) = 4.75 mgd Actual flow rate (ave daily) = 2.2 mgd Influent BOD₅ concentration = 170 mg/L Influent TSS concentration = 260 mg/L Influent TKN concentration = 30 mg/L Volume of the ditch = 2.0 mil gal

Pell City, AL WWTP



Pell City, AL WWTP



Previous Operating Conditions at Pell City, AL WWTP

Actual aeration basin detention time = 22 hrs F/M ratio = 0.045 lb BOD₅/(day-lb MLVSS) VOLR = 11.7 lb BOD₅/(day-thou cu ft) Solids Retention Time (SRT) = 50 days MLSS concentration = 5,800 mg/L MLVSS concentration = 4,100 mg/L TSS sludge production = 1,770 lb/day Oxygen requirements = 6,200 lb/day

Previous Operating Conditions at Pell City, AL WWTP

Oxygen supplied = 6,000 lb/day (three 50-hp rotors running 24/7)

Effluent $CBOD_5$ conc = 4 mg/L

Effluent TSS conc = 8 mg/L

Effluent amm-N conc = 0.2 mg/L

Effluent TKN conc = 1 mg/L

Effluent Total P conc = 0.7 mg/L (w/ alum)

Effluent NO_x conc = 15 mg/L

Effluent Total N conc = 16 mg/L

Denitrification at Pell City, AL WWTP

Turn three 50-hp rotors off 5 hours per day

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Estimated energy savings = 14,400 kWh/mo
Estimated cost savings = $1,700/mo (12%)
Current effluent NO<sub>x</sub> conc = 10 mg/L
Current effluent Total N conc = 11 mg/L
Reduction in effluent Total N = 92 lb/day
= 17 tons/year
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